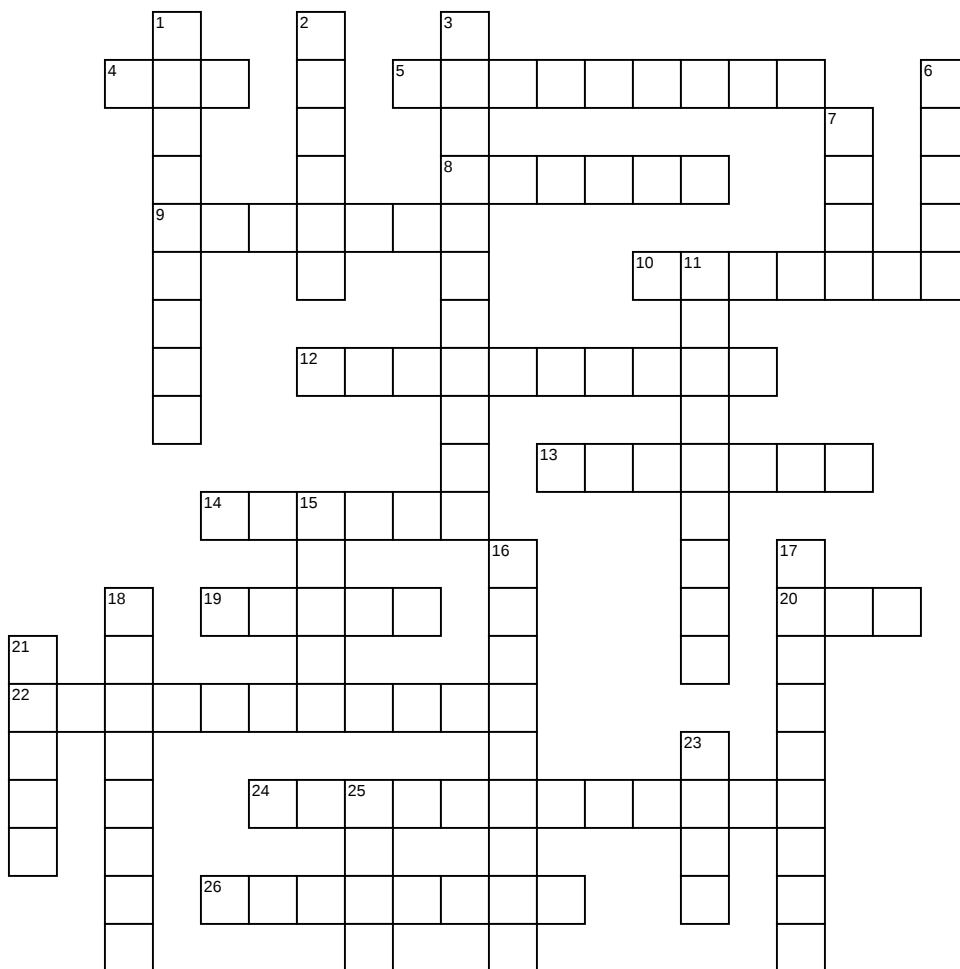


NAME: _____ DATE: _____ PERIOD: _____

Chemistry Activity – Naming Ionic Compounds – Form 1 ©



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ACROSS

- 4 barium oxide
- 5 ammonium phosphate
- 8 copper(I) persulfate
- 9 strontium hydroxide
- 10 sodium oxalate
- 12 copper(II) sulfate pentahydrate
- 13 gold(I) thiosulfate
- 14 rubidium perchlorate
- 19 iron(III) borate
- 20 nickel(III) phosphide
- 22 magnesium acetate
- 24 silver hydrogen chromate tetrahydrate
- 26 gallium cyanate

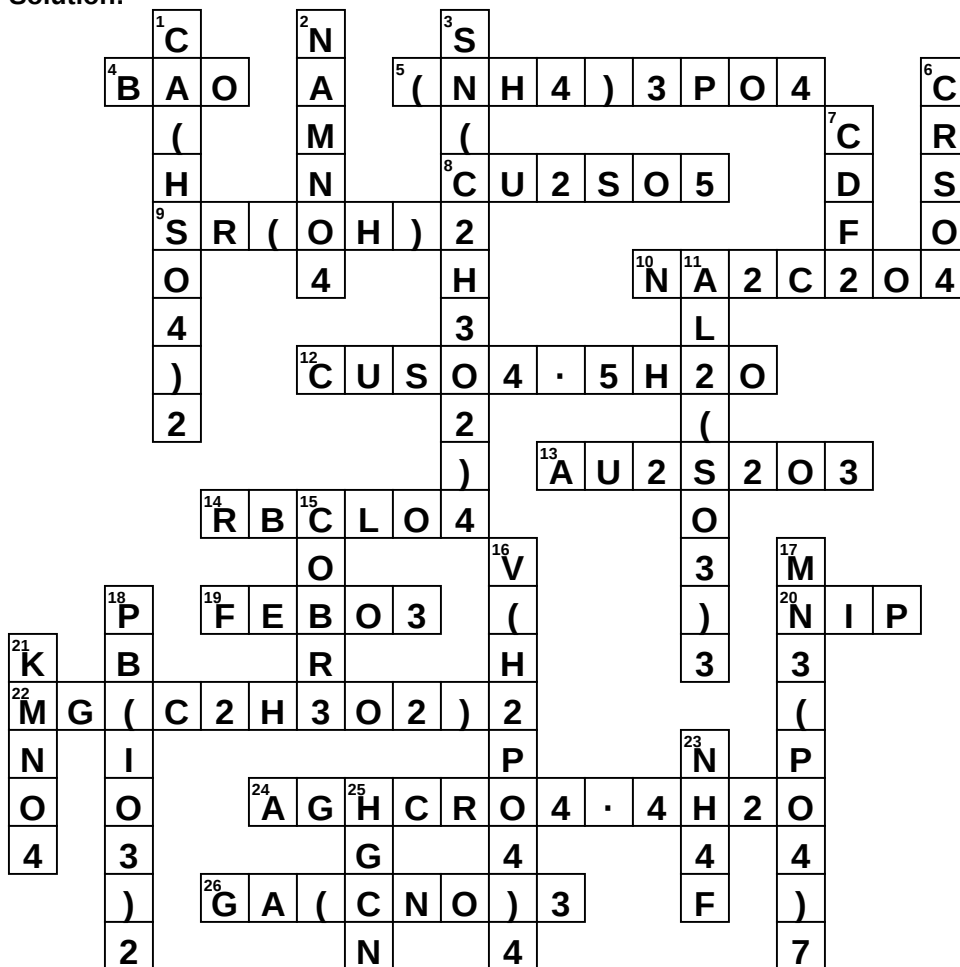
DOWN

- 1 calcium bisulfate
- 2 sodium permanganate
- 3 tin(IV) acetate
- 6 chromium(II) sulfate
- 7 cadmium fluoride
- 11 Aluminum Sulfite
- 15 cobalt(III) bromide
- 16 vanadium(IV) dihydrogen phosphate
- 17 manganese(VII) phosphate
- 18 lead(II) iodate
- 21 potassium permanganate
- 23 ammonium fluoride
- 25 mercury(I) cyanide

NAME: _____ DATE: _____ PERIOD: _____

Chemistry Activity – Naming Ionic Compounds – Form 1 ©

Solution:



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